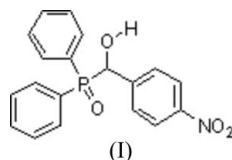
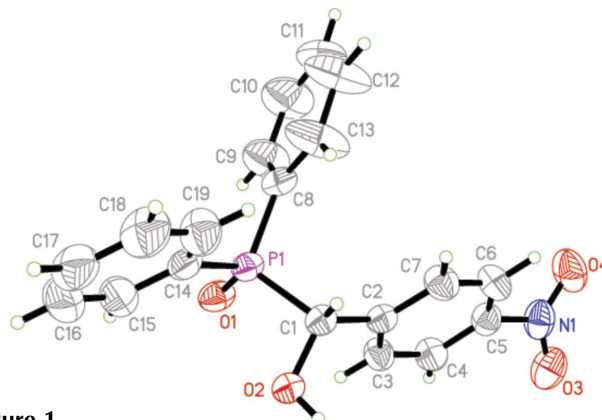


Wan-Yun Liu,<sup>a,‡</sup> Ping Huo,<sup>a</sup>  
Yu-Xing Gao,<sup>b</sup> Peng Liu<sup>b</sup> and  
Yu-Fen Zhao<sup>b,\*</sup><sup>a</sup>Department of Chemistry and Bioengineering,  
Yichun University, Yichun 336000, People's  
Republic of China, and <sup>b</sup>Department of  
Chemistry, The Key Laboratory for Chemical  
Biology of Fujian Province, College of Chemistry  
and Chemical Engineering, Xiamen University,  
Xiamen 361005, People's Republic of China‡ Current address: Department of Chemistry,  
The Key Laboratory for Chemical Biology of  
Fujian Province, College of Chemistry and  
Chemical Engineering, Xiamen University,  
Xiamen 361005, People's Republic of China

Correspondence e-mail: yfzhao@xmu.edu.cn

**Key indicators**Single-crystal X-ray study  
 $T = 298\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$   
 $R$  factor = 0.042  
 $wR$  factor = 0.128  
Data-to-parameter ratio = 14.3For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.**(Diphenylphosphoryl)(4-nitrophenyl)methanol**The title compound,  $\text{C}_{19}\text{H}_{16}\text{NO}_4\text{P}$ , was obtained by the  
reaction of diphenylphosphine oxide and 4-nitrobenzal-  
dehyde. Molecules are linked into chains, running along the  $b$   
axis, by  $\text{O}-\text{H}\cdots\text{O}$  and  $\text{C}-\text{H}\cdots\text{O}$  intermolecular hydrogen  
bonds.Received 20 January 2007  
Accepted 24 January 2007**Comment**The title compound, (I), is an analogue of (diphenyl-  
phosphinoyl)phenylmethanol, which is employed as a ligand  
in the rhodium-catalysed hydroformylation of alkenes, with  
good yields and regioselectivities (Clark *et al.*, 2002).The molecular structure of (I) is shown in Fig. 1. Bond  
lengths and angles in (I) are in agreement with those reported  
for similar compounds (Dankowski *et al.*, 1979; Fang *et al.*,  
2006). The dihedral angle between the C8-phenyl and C14-  
phenyl planes is  $56.1(1)^\circ$ . The  $\text{O3}-\text{N1}-\text{C5}-\text{C4}$  and  $\text{O4}-$   
 $\text{N1}-\text{C5}-\text{C4}$  torsion angles of  $-35.0(2)$  and  $144.94(18)^\circ$ ,  
respectively, indicate that the nitro group is twisted away from  
the attached ring. $\text{O}-\text{H}\cdots\text{O}$  and  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds (Table 1)  
involving the hydroxyl group link the molecules into a chain  
running along the  $b$  axis. In addition,  $\text{O3}\cdots\text{N1}(2-x, -y, -z)$   
[2.849(3) Å],  $\text{O3}\cdots\text{O3}(2-x, -y, -z)$  [2.896(3) Å] and  
 $\text{O4}\cdots\text{O4}(2-x, 1-y, -z)$  [2.892(3) Å] short contacts are  
observed in the crystal structure.**Figure 1**  
The molecular structure of (I), showing 50% probability displacement  
ellipsoids (arbitrary spheres for H atoms).

## Experimental

To a precooled solution of 4-nitrobenzaldehyde (0.15 g, 1.0 mmol) and diphenylphosphine oxide (0.20 g, 1.0 mmol) in tetrahydrofuran (10 ml) at 273 K was added dropwise triethylamine (0.15 ml, 1.0 mmol). The cooling bath was removed and the mixture warmed to ambient temperature for 3 h. The solvent was concentrated under vacuum and the crude product was purified by column chromatography (petroleum ether–ethyl acetate, 1:1) to give the title compound as a white solid in 88% yield. Single crystals of (I) were obtained by slow evaporation of a methanol solution.

### Crystal data

$C_{19}H_{16}NO_4P$   
 $M_r = 353.30$   
 Monoclinic,  $P2_1/n$   
 $a = 11.449$  (3) Å  
 $b = 6.1417$  (15) Å  
 $c = 23.624$  (6) Å  
 $\beta = 94.409$  (5)°  
 $V = 1656.2$  (7) Å<sup>3</sup>

$Z = 4$   
 $D_x = 1.417$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 $\mu = 0.19$  mm<sup>-1</sup>  
 $T = 298$  (2) K  
 Needle, colourless  
 $0.72 \times 0.27 \times 0.20$  mm

### Data collection

Bruker APEX area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Bruker, 2001)  
 $T_{\min} = 0.875$ ,  $T_{\max} = 0.963$

8522 measured reflections  
 3236 independent reflections  
 2800 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.020$   
 $\theta_{\max} = 26.0^\circ$

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.042$   
 $wR(F^2) = 0.128$   
 $S = 1.01$   
 3236 reflections  
 226 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0846P)^2 + 0.3649P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.35$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.20$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O2-H2A\cdots O1^i$	0.82	1.93	2.696 (2)	156
$C1-H1B\cdots O2^i$	0.98	2.47	3.050 (2)	117

Symmetry code: (i)  $-x + \frac{3}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$ .

All H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C–H = 0.93 (aromatic) or 0.98 Å (methine), O–H = 0.82 Å, and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  and  $1.5U_{\text{eq}}(\text{O})$ .

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: SHELXL97.

The authors thank the Key Project of the Chinese Ministry of Education (Key grant No. 104201) for supporting this work, and Mr. R.-B. Huang for technical assistance.

## References

- Bruker (2001). SAINT (Version 6.22), SMART (Version 5.625) and SADABS (Version 2.03). Bruker AXS Inc., Madison, Wisconsin, USA.
- Clark, H. J., Wang, R. & Alper, H. (2002). *J. Org. Chem.* **67**, 6224–6225.
- Dankowski, M., Praefcke, K., Nyburg, S. C. & Wong-Ng, W. (1979). *Phosphorus Sulfur*, **7**, 275–279.
- Fang, M.-J., Fang, H., Zeng, Z.-P., Luo, S.-N. & Zhao, Y.-F. (2006). *Acta Cryst.* **E62**, o1998–o1999.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.